

(19)

Europäisches Patentamt

European Patent Office

Office européen des brevets



(11)

EP 0 943 587 A1

(12)

EUROPEAN PATENT APPLICATION

(43) Date of publication:
22.09.1999 Bulletin 1999/38(51) Int. Cl.⁶: C03C 17/00, C03C 17/22,
C03C 17/34, C03C 17/42,
C09D 1/00, C09D 5/00

(21) Application number: 98308885.7

(22) Date of filing: 29.10.1998

(84) Designated Contracting States:
AT BE CH CY DE DK ES FI FR GB GR IE IT LI LU
MC NL PT SE
Designated Extension States:
AL LT LV MK RO SI

(30) Priority: 16.03.1998 JP 6491698

(71) Applicant:
SUMITOMO METAL MINING COMPANY LIMITED
Tokyo 105 (JP)(72) Inventors:

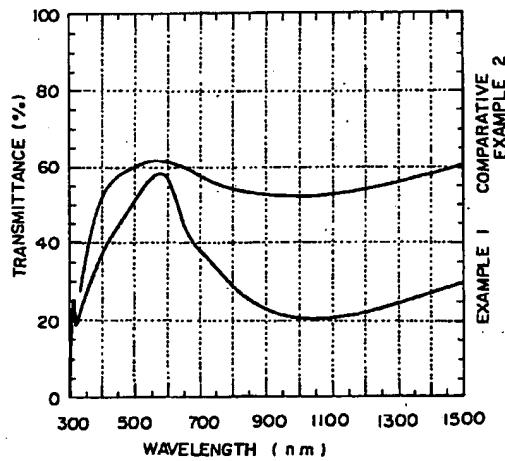
- Kuno, Hiroko
Matsudo-shi, Chiba Prefecture (JP)
- Takeda, Hiromitsu
Ichikawa-shi, Chiba Prefecture (JP)
- Adachi, Kenji
Inzai-shi, Chiba Prefecture (JP)

(74) Representative:
Harrison, Ivor Stanley et al
Withers & Rogers,
Goldings House, 2 Hays Lane
London SE1 2HW (GB)

(54) Near Infra-red shielding single or multilayer film and a coating liquid for forming the same

(57) A film is provided for cutting off heat rays, the film having a high transmittance and a low reflectivity of visible light, a low transmittance and a high reflectivity of near infrared radiation and a surface resistivity of at least about 10^6 ohms/square. The film is easy to form at a low cost by coating a surface with a coating liquid and heating it. The coating liquid is a dispersion containing fine particles of borides having an average diameter not exceeding 100 nm. The dispersion may further contain fine particles of at least one ruthenium or iridium oxide having an average diameter not exceeding 100 nm.

Fig. 1



EP 0 943 587 A1

Description**BACKGROUND OF THE INVENTION**5 **1. Field of the Invention:**

[0001] This invention relates to a film for cutting off the passage of heat rays through a transparent or semi-transparent material, such as glass or plastics, for e.g. the windows of vehicles, buildings, offices or houses, telephone booths, show windows, or lamps, and a coating liquid used for forming the same.

10 **2. Description of the Prior Art:**

[0002] A sheet of glass having a film formed on its surface from a material reflecting visible and infrared wavelengths has been used as heat-ray reflecting glass for removing or reducing heat emitted by the sun, or an electric bulb, or any other source of light. The film has been formed from a material selected from among metal oxides such as FeOx, CoOx, CrOx and TiOx, and metals such as Ag, Au, Cu, Ni and Al.

[0003] These materials, however, not only reflect or absorb near-infrared radiation known as a source of heat, but also have a low transmittance of visible light. A transparent material used for buildings, vehicles, telephone booths, etc. are required to have a high transmittance of visible light. If any of the known materials as mentioned above is used for any such purpose, it has been necessary to form a very thin film, and it has been usual practice to employ a process, such as spraying and baking, CVD, sputtering or vacuum vapor deposition, to form a film having a very small thickness in the order of 10 nm.

[0004] These processes have, however, been unsatisfactory for a number of reasons including the necessity for a large apparatus, or vacuum equipment, low productivity, difficulty in forming a film covering a large area, and the expenditure of a large amount of cost.

[0005] Referring to the drawbacks of the films formed from the known materials, a film having a small thickness and thereby a high transmittance of light has a low property of cutting off heat rays, while a film is darkened if its thickness is sufficiently large to give it a high property of cutting off heat rays. A film having a high property of cutting off heat rays is also likely to have a high reflectivity of visible light and present a surface shining like a mirror and spoiled in appearance. Moreover, many of these materials give a film having a high electric conductivity which reflects radio waves and makes them unreceivable by a radio or television receiver, or a portable telephone, or causes the disturbance of waves in or around the structure in which the film is used.

[0006] These drawbacks can be overcome by a film having a low reflectivity of visible light and a high absorption or reflectivity of near infrared radiation, as well as a surface resistivity of or above about $10^6 \Omega/\square$. There has, however, not been known any such film, or any material suitable therefor.

[0007] Antimony tin oxide (ATO) and indium tin oxide (ITO) are known as materials having a high transmittance of visible light and a power of cutting off heat rays. These materials have a relatively low reflectivity of visible light and does not give a shining surface to a film, but as their plasma wavelengths fall in relatively long wavelengths in the near infrared region, a film formed from either material has been unsatisfactory in its property of reflecting or absorbing near infrared radiation close to visible light. Moreover, the film has been so high in electric conductivity as to reflect radio waves.

SUMMARY OF THE INVENTION

[0008] It is, therefore, an object of this invention to provide a film for cutting off heat rays which has a high transmittance and a low reflectivity of visible light, a low transmittance of near infrared radiation and a surface resistivity of or above about $10^6 \Omega/\square$. It is another object of this invention to provide a coating liquid which makes it possible to form any such film easily at a low cost even on a large surface.

[0009] We, the inventors of this invention, considered that it would be effective to use borides having a large quantity of free electrons, and as a result of our extensive research work, we have found as a basis for our invention that a film of high density formed from a dispersion of ultrafine particles of any such material has a maximum transmittance of light in the visible region, and a minimum transmittance in the near infrared region close to visible light.

[0010] According to this invention, there is provided a coating liquid which comprises a dispersion of fine particles of a boride of one or more metals selected from the group consisting of La, Pr, Nd, Ce, Gd, Tb, Dy, Ho, Y, Sm, Eu, Er, Tm, Yb, Lu, Sr, Ca, Ti, Zr, Hf, V, Nb, Ta, Cr, Mo and W, those particles having an average diameter not exceeding 100 nm.

[0011] The dispersion may also contain one or both of fine particles of ruthenium oxide and fine particles of iridium oxide, both having an average diameter not exceeding 100 nm.

[0012] In either event, the dispersion may also contain one or more of alkoxides of silicon, zirconium, titanium and aluminum, and partially polymerised products of those alkoxides.

[0013] The dispersion may further contain a resinous binder.

[0014] According to this invention, there is also provided a film formed on a base for cutting off heat rays by coating it with a coating liquid and heating it, the coating liquid comprising a dispersion prepared by dispersing fine particles of a boride of one or more metals selected from among La, Pr, Nd, Ce, Gd, Tb, Dy, Ho, Y, Sm, Eu, Er, Tm, Yb, Lu, Sr, Ca, Ti, Zr, Hf, V, Nb, Ta, Cr, Mo and W in a resinous binder, or a binder containing one or more metal oxides selected from among silicon, zirconium, titanium and aluminum oxides.

[0015] The film may be covered with an oxide film containing one or more metal oxides selected from among silicon, zirconium, titanium and aluminum oxides, or a resin film to form a multilayer film for cutting off heat rays.

[0016] In either event, the film has a maximum transmittance of light at a wavelength of 400 to 700 nm and a minimum transmittance at a wavelength of 700 to 1800 nm, and its maximum and minimum transmittances have a difference of 15 points or more in percentage. It also has a surface resistivity of $10^6 \Omega/\square$ or above. Therefore, the film of this invention does not have a shining surface, and has a high property of transmitting radio waves.

BRIEF DESCRIPTION OF THE DRAWING

15

[0017]

Figure 1 is a graph showing the transmittance of the films according to Example 1 and Comparative Example 2 which will hereinafter be described. Figure 2 is a graph showing the transmittance of the film according to Example 20 8.

DETAILED DESCRIPTION OF THE INVENTION

[0018] The coating liquid of this invention contains fine particles of a metal boride, or borides, such as lanthanum boride (LaB_6), praseodymium boride (PrB_6), neodymium boride (NdB_6), cerium boride (CeB_6), gadolinium boride (GdB_6), terbium boride (TbB_6), dysprorium boride (DyB_6), holmium boride (HoB_6), yttrium boride (YB_6), samarium boride (SmB_6), europium boride (EuB_6), erbium boride (ErB_6), thulium boride (TmB_6), ytterbium boride (YbB_6), lutetium boride (LuB_6), strontium boride (SrB_6), calcium boride (CaB_6), titanium boride (TiB_2), zirconium boride (ZrB_2), hafnium boride (HfB_2), vanadium boride (VB_2), tantalum boride (TaB_2), chromium borides (CrB and CrB_2), molybdenum borides (MoB_2 , Mo_2B_5 and MoB) and tungsten boride (W_2B_5).

[0019] Although it is preferable to use boride particles not having oxidized surfaces, it is usually the case that boride particles have slightly oxidized surfaces, or some oxidation may be unavoidable during the step of preparing a dispersion. In any such event, they are equally effective for cutting off heat rays. Although boride particles having a higher crystallinity may produce a greater effect of cutting off heat rays, even particles having so low a crystallinity as to reveal a very broad X-ray diffraction peak are effective for cutting off heat rays if the basic atomic bonds in the particles consist of boron and a metal.

[0020] The coating liquid of this invention may further contain fine particles of one or more ruthenium and iridium oxides, such as ruthenium dioxide (RuO_2), lead ruthenate ($\text{Pb}_2\text{Ru}_2\text{O}_{6.5}$), bismuth ruthenate ($\text{Bi}_2\text{Ru}_2\text{O}_7$), iridium dioxide (IrO_2), bismuth iridate ($\text{Bi}_2\text{Ir}_2\text{O}_7$) and lead iridate ($\text{Pb}_2\text{Ir}_2\text{O}_{6.5}$). These particles are stable as oxides, and have a large quantity of free electrons, and are, therefore, very effective for cutting off heat rays.

[0021] These boride and oxide particles have a grayish, brownish or greenish black color, but if their diameters are sufficiently small as compared with the visible wavelengths, a thin film in which they are dispersed transmits visible light. It, however, has a strong power of cutting off infrared radiation for some unclear reason. Probably because those particles have a large quantity of free electrons, they reflect or absorb heat rays selectively, as the free electron plasmons in and on the particles have a plasma frequency in the near infrared region close to visible light.

[0022] Experiments show that a film in which those particles having a sufficiently small diameter are uniformly dispersed has a maximum transmittance of light in a wavelength range of 400 to 700 nm and a minimum transmittance of light in a wavelength range of 700 to 1800 nm, and that its maximum and minimum transmittances have a difference of 15 points or more in percentage. Insofar as visible light has a wavelength of 380 to 780 nm, and as sensitivity of human eyes is expressed by a bell-shaped curve having a peak in the vicinity of 550 nm, it is obvious that such a film effectively transmits visible light and reflects or absorbs heat rays.

[0023] The boride and oxide particles preferably have an average diameter not exceeding 100 nm. If their average diameter exceeds 100 nm, they fail to form a film having a profile of transmittance as stated above, i.e., a profile having a maximum transmittance of light in a wavelength range of 400 to 700 nm and a minimum transmittance of light in a wavelength range of 700 to 1800 nm, with a difference of 15 points or more in percentage, is not obtained, and yields a grayish film having a monotonously decreased transmittance. Particles having an average diameter exceeding 100 nm are also likely to cohere or agglomerate and thereby tend to settle in the coating liquid. They, or coarse particles formed by their cohesion are also undesirable, since they serve as a source of scattering light and form a film having a

haze, or lower its transmittance of visible light. Particles having a minimum diameter of, say, 2 nm are already economically available, though it is not the minimum diameter of the particles to be used for the purpose of this invention.

[0024] The particles may be dispersed in any medium that is suitable under the coating conditions or environment, or in view of the alkoxide, or synthetic resin binder employed. The dispersing medium may, for example, be water, or any of the organic solvents such as alcohols, ethers, esters or ketones, and an acid or alkali may be added to adjust the pH of the dispersion. The dispersion may further contain a surface active agent, or coupling agent which can further improve the stability of the particles dispersed therein. The amount of each additive may be up to 30%, and preferably up to 5%, by weight of the inorganic particles.

[0025] While the conduction of electricity in a film formed from the coating liquid takes place along a path formed by the particles contacting one another, it is possible to cut the path partly by varying, for example, the amount of the surface active, or coupling agent in the coating liquid, and it is thereby easy to lower the electric conductivity of the film so that it may have a surface resistivity of $10^6 \Omega/\square$ or above. If the coating liquid further contains an alkoxide of silicon, zirconium, titanium or aluminum, or a partially polymerized product thereof, or a synthetic resin binder, the amount thereof can be altered to form a film having an adequately controlled electric conductivity.

[0026] The dispersion may be prepared by any method that can form a uniform dispersion, or by employing, for example, a bead, ball or sand mill, or an ultrasonic disperser.

[0027] The film of this invention is formed by the deposition of the boride (and oxide) particles with a high density on a substrate. If the coating liquid contains any alkoxide of silicon, zirconium, titanium or aluminum, any partially polymerized product thereof, or any synthetic resin binder, it is effective for improving the adhesion of the particles to the substrate and the hardness of the film. If the film is covered with a second layer containing an alkoxide of silicon, zirconium, titanium or aluminum, a partially polymerized product thereof, or a synthetic resin, it is possible to improve the adhesion of the film to the substrate and its hardness and weatherability to a still further extent.

[0028] If the coating liquid does not contain any alkoxide of silicon, zirconium, titanium or aluminum, any partially polymerized product thereof, or any synthetic resin binder, the film is composed solely of the boride (and oxide) particles deposited on the substrate. Although it is itself effective for cutting off heat rays, it is possible to reduce its haze and thereby improve its transmittance of visible light, as well as improving its adhesion to the substrate, if a second layer is formed on the film as a first layer by a second coating liquid containing an alkoxide of silicon, zirconium, titanium or aluminum, a partially polymerized product thereof, or a synthetic resin binder to form a multilayer film, since the second coating liquid fills the interstices of the particles in the first layer.

[0029] The second layer by which the first layer of the boride (and oxide) particles is firmly bonded to the substrate is easy to form at a low cost by coating, though it can also be formed by sputtering or vapor deposition. The second coating liquid may contain one or more of the alkoxides of silicon, zirconium, titanium and aluminum, and their partially polymerized products in water or alcohol, and preferably has an alkoxide, etc. content not exceeding 40% by weight in terms of the oxide formed by heating. An acid or alkali may be added to the coating liquid to adjust its pH as required.

If the substrate is heated after the coating liquid has been applied onto the first layer to form the second layer, it is easy to form a film of an oxide of silicon, zirconium, titanium or aluminum.

[0030] Any method, such as spin, spray, dip, roll or flow coating, or screen printing, can be employed to apply the coating liquids if it can form a smooth and uniform film having a small thickness.

[0031] After the application of the second coating liquid, the substrate is preferably heated at a temperature of 100°C or above, and more preferably at or above the boiling point of the solvent in the coating liquid. If its heating temperature is lower than 100°C, there is every likelihood that the alkoxide, or hydrolyzed polymerization product thereof may remain incompletely polymerized in the film, or that the solvent, such as water or an organic solvent, may remain unremoved in the film and lower its transmittance of visible light. If the coating liquid contains a synthetic resin binder, a method which is suitable for curing the resin may be applied to the layer. If it contains, for example, an ultraviolet-curing resin, the layer may be exposed to ultraviolet radiation, or if it contains a cold-setting resin, it is left to set. Thus, the coating liquid is applicable to any existing window glass, etc. at the site of the job, and is useful for a wide range of purposes.

[0032] An organosilazane solution can be used as a binder in the coating liquid of this invention, or as the second coating, or overcoating liquid. There is commercially available an organosilazane solution which has a polymerizing and curing temperature of 100°C or below as a result of the modification of the group in the side chain, or the addition of an oxidizing catalyst, and which can be used to form a film at a considerably low temperature. A commercially available organo-silicate solutions can be used as a binder which is curable at room temperature. In either event, an inorganic film of SiO₂ is formed upon curing and is superior to a resin film in weatherability and strength.

[0033] The film of this invention in which the ultrafine particles are dispersed has a great advantage over any film formed by any physical film-forming process as mentioned before, since it does not reflect visible light so much as to present a shining surface, and exhibits a high absorption in the near infrared region. If it is desirable to restrain the reflection of visible light to a further extent, it is easily possible to form a multilayer film having a visible light reflectivity not exceeding 1% if a film having a low refractive index, such as of SiO₂ or MgF, is formed on the film of this invention.

[0034] The coating liquid of this invention may further contain ultrafine particles of e.g. ATO, ITO or aluminum-contain-

ing zinc oxide to form a film having a still higher transmittance of visible light. The addition of these transparent ultrafine particles in a substantial amount makes it possible to form a film having a high absorbance of near infrared radiation close in wavelength to visible light and thereby a high transmittance of visible light. If the coating liquid of this invention is added to a dispersion of those particles, on the other hand, it colors a film of those particles and adds to its power of cutting off heat rays, even if it may be added only in a small amount, and its addition, therefore, makes it possible to reduce the use of ITO, etc. greatly and thereby prepare a dispersion thereof at a lower cost.

[0035] The coating liquid of this invention may also contain fine particles of one or more other substances including inorganic substances such as titanium, zinc and cerium oxides, and organic substances such as benzophenone and benzotriazole, to form a film having an improved power of cutting off ultraviolet radiation which is harmful to the human body, as well as an improved power of cutting off infrared radiation.

[0036] The coating liquid of this invention makes it possible to form a highly stable film of uniform thickness, since it is a dispersion of the inorganic fine particles as described, and does not rely upon the decomposition or chemical reaction of its components by the heat of baking for forming a film. A dispersion of an appropriate mixture of particles of inorganic substances makes it possible to form a film having a high power of cutting off heat rays. The film hardly undergoes any deterioration in color or function, even if it may, for example, be exposed to sunlight (among others, ultraviolet radiation), since the inorganic substances are by far higher in weatherability than the organic substances.

[0037] The invention will now be described more specifically based on examples embodying it and comparative examples.

20 EXAMPLE 1

[0038] 8 g of fine particles of LaB_6 having an average diameter of 67 nm, 80 g of diacetone alcohol (DAA) and appropriate amounts of water and a dispersant were mixed for 100 hours in a ball mill containing zirconia balls having a diameter of 4 mm to prepare 100 g of a dispersion of LaB_6 (dispersion A). An ethyl silicate solution was prepared by mixing 6 g of Ethyl Silicate 40, product of Tama Chemical Industrial Co., Ltd. having an average polymerization degree of 4 to 5, 31 g of ethanol, 8 g of a 5% aqueous solution of hydrochloric acid and 5 g of water, and 50 g thereof, 800 g of water and 300 g of ethanol were thoroughly mixed and stirred to prepare 1,150 g of a mixed ethyl silicate solution (solution B).

[0039] The dispersion A and the solution B were mixed and stirred in such proportions as to yield a coating liquid (liquid C) having a LaB_6 content of 1.0% and a $\text{LaB}_6/\text{SiO}_2$ ratio of 4/1. 15 g of liquid C was dropped from a beaker onto a sheet of soda-lime glass employed as a substrate, measuring 200 mm square by 3 mm thick and rotating at a speed of 145 rpm, and after three minutes, its rotation was stopped. The substrate was heated at 180°C for 30 minutes in an electric furnace to have a film formed thereon as intended.

[0040] The spectral characteristics of the film were examined by using a Hitachi Limited spectrophotometer. Figure 1 shows a profile of transmittance of light as exhibited by the film. It exhibited a maximum transmittance at a wavelength in the vicinity of 580 nm and a minimum transmittance in the vicinity of 1055 nm, and the values of its maximum and minimum transmittance had a difference of as large as 38 points. The calculation made in accordance with JIS-R-3106 gave a visible light transmittance of 54.6% as the net value for the film.

[0041] The light transmitted through the film presented a beautiful dark green color. The film had a visible light reflectivity of as low as 2.2% and did not present at all any shining surface like that of commercially available heat-ray reflecting glass.

[0042] The film had a surface resistivity of $3.2 \times 10^{11} \Omega/\square$ as measured by a Mitsubishi Chemical surface resistivity meter. It was sufficiently high to transmit radio waves quite satisfactorily.

45 COMPARATIVE EXAMPLE 1

[0043] Commercially available heat-ray reflecting bronze glass, on which a heat-ray reflecting film had been formed by an expensive physical film-forming process as mentioned before, was examined for its spectral transmittance at wavelengths of 340 to 1800 nm, and its transmittance of visible light was determined as 38.8% by calculation in accordance with JIS-R-3106. It had a visible light reflectivity of as high as 34.2% and presented a surface shining like a mirror. It had a surface resistivity of $83 \Omega/\square$ which was obviously too low for the film to transmit radio waves satisfactorily.

EXAMPLE 2

[0044] The liquid C as obtained in Example 1 was used to form a first layer on sheet glass by spin coating, and after its spinning was continued for three minutes, 15 g of a silicate solution obtained by diluting the solution B with ethanol, and having a solid SiO_2 content of 0.9% was dropped from a beaker onto the glass to form a second layer, and after three minutes, its spinning was stopped. Then, the glass substrate was heated at 180°C for 30 minutes in an electric furnace to have a two-layer film formed thereon as intended.

[0045] Example 1 was repeated for examining the spectral characteristics of the film. It had a transmittance of visible light raised to 56.2% and a reflectivity thereof lowered to 1.7%, as compared with the results of Example 1. The film had its reflectivity of visible light measured again after having a black tape bonded to its rear surface to eliminate any reflection of light thereon. It showed a reflectivity of 0.5% and presented a surface close to that of non-reflective glass.

- 5 [0046] It showed a maximum and a minimum transmittance substantially at the same wavelengths with the single-layer film according to Example 1, and was obviously comparable to it in its power of cutting off heat rays.

EXAMPLE 3

- 10 [0047] 8 g of fine particles of LaB_6 having an average diameter of 67 nm, 80 g of isophorone and appropriate amounts of water and a dispersant were mixed by zirconia balls in a ball mill for 100 hours to prepare 100 g of an isophorone dispersion of LaB_6 (dispersion D). An epoxy resin was dissolved in isophorone to prepare a binder solution containing 50% by weight of epoxy resin (solution E). The dispersion D, solution E and ethanol were strongly mixed and stirred to prepare a coating liquid having a solid LaB_6 and epoxy resin content of 1.4% by weight and a LaB_6 /epoxy resin ratio by weight of 70/30. Example 1 was repeated for forming a film.
- 15

EXAMPLE 4

- 20 [0048] A coating liquid was prepared by using a Shinetsu Silicone cold-setting silicate solution, X-40-9740, as a binder instead of solution B and otherwise repeating Example 1. A film was formed by leaving the coating liquid at a room temperature of 25°C for two days to dry on a substrate instead of heating it, and otherwise repeating Example 1.

EXAMPLE 5

- 25 [0049] A coating liquid having a LaB_6 content of 1.0% and a $\text{LaB}_6/\text{SiO}_2$ ratio of 4/1 was prepared by using a cold-setting polyperhydrosilazane solution of NE Chemcat Co., Ltd. as a binder instead of solution E, mixing and stirring it, the dispersion D and xylene, and otherwise repeating Example 3. A film was formed by heating the substrate at 80°C and otherwise repeating Example 1.

30 EXAMPLE 6

[0050] A coating liquid was prepared by using fine particles of PrB_6 having an average diameter of 75 nm instead of LaB_6 to prepare the dispersion A and otherwise repeating Example 1, and was used to form a film by repeating Example 1.

- 35 Example 7-21

[0051] A coating liquid was prepared by using fine particles of \square having an average diameter of \square nm instead of LaB_6 to prepare the dispersion A and otherwise repeating Example 1, and was used to form a film by repeating Example 1.

40

45

50

55

	X	Y
Example 7	NdB_6	49
Example 8	CeB_6	61
Example 9	GdB_6	30
Example 10	TbB_6	67
Example 11	DyB_6	83
Example 12	HoB_6	77
Example 13	YB_6	48
Example 14	SmB_6	69
Example 15	EuB_6	41
Example 16	ErB_6	35

(continued)

	X	Y
Example 17	TmB ₆	56
Example 18	YbB ₆	82
Example 19	LuB ₆	53
Example 20	SrB ₆	68
Example 21	CaB ₆	82

5

10

EXAMPLE 22

[0052] A coating liquid was prepared by using fine particles of TiB₂ having an average diameter of 84 nm instead of LaB₆ to prepare the dispersion A and otherwise repeating Example 1, and was used to form a film by repeating Example 1.

EXAMPLE 23

[0053] A coating liquid was prepared by using fine particles of TaB₂ having an average diameter of 63 nm instead of LaB₆ to prepare the dispersion A and otherwise repeating Example 1, and was used to form a film by repeating Example 1.

EXAMPLE 24

25

[0054] A coating liquid was prepared by using fine particles of MoB having an average diameter of 59 nm instead of LaB₆ to prepare the dispersion A and otherwise repeating Example 1, and was used to form a film by repeating Example 1.

EXAMPLE 25

[0055] A coating liquid was prepared by using fine particles of W₂B₅ having an average diameter of 85 nm instead of LaB₆ to prepare the dispersion A and otherwise repeating Example 1, and was used to form a film by repeating Example 1.

35

EXAMPLE 26

[0056] 15 g of fine particles of ruthenium oxide (RuO₂) having an average diameter of 30 nm, 23 g of N-methyl-2-pyrrolidone (NMP), 57 g of diacetone alcohol and appropriate amounts of water and a dispersant were mixed for 100 hours in a ball mill by using zirconia balls having a diameter of 4 mm to prepare 100 g of a dispersion of RuO₂. This dispersion and the silicate solution B (Example 1) were mixed and stirred to prepare a silicate solution (solution F) containing RuO₂ dispersed therein at a concentration of 1% and having a RuO₂/SiO₂ ratio of 4/1. The dispersion A (Example 1) and the solution F were thoroughly mixed and stirred to prepare a coating liquid containing RuO₂ and LaB₆ at a ratio by weight of 1 : 1. The coating liquid was used to form a film by repeating Example 1.

45

EXAMPLE 27

[0057] A coating liquid containing a mixture of IrO₂ and LaB₆ dispersed in a silicate solution was prepared by using fine particles of IrO₂ having an average diameter of 28 nm and otherwise repeating Example 11, and was used to form a film.

[0058] In Examples 1 to 12, as well as Comparative Examples 1 to 3, the transmittance and surface resistivity of the film were determined by repeating Example 1. The results, including those of Examples 1 and 2, are shown in Table 1 below. As is obvious from Table 1, all of the films according to Examples 1 to 12 of this invention have a maximum transmittance of visible light at a wavelength of 400 to 700 nm and a minimum transmittance at a wavelength of 700 to 1800 nm with a difference of more than 15 points in percentage therebetween, and are useful for cutting off heat rays. They all have a reflectivity of visible light not exceeding 8% and present a surface not shining like a mirror. Moreover, they all have a surface resistivity higher than $8 \times 10^{10} \Omega/\square$ and transmit radio waves satisfactorily.

EXAMPLE 28

[0059] 35g of fine particles of ITO having an average diameter of 55 nm, 61 g of diacetone alcohol (DAA) and 4.0 g of a coupling agent for dispersing the fine particles were mixed for twelve hours in a ball mill by using zirconia balls in a ball mill to prepare 100 g of a dispersant of ITO fine particles (solution C). The solution C, the dispersion A disclosed in Example 1 and silicon based UV curing resin produced by Shin-Etsu Silicon Co., Ltd. were thoroughly mixed while diluting with ethanol to prepare a solution having the composition as shown by Example 20 in Table 1. 15 g of the thus prepared solution was dropped from a beaker onto a sheet of soda-glass employed as a substrate, measuring 200 mm square by 3 mm thick and rotating at a speed of 200 rpm and after five minutes, its rotation was stopped. The resultant substrate was heated at 100°C for 2 minutes in an electric furnace to evaporate the solvent and, then, subjected to a ultraviolet irradiation for two minutes by using a high-pressure mercury lamp to obtain an aimed film. The optical characteristics of the thus obtained film are shown in Table 1.

Example 29

[0060] 35 g of fine particles of ATO having an average diameter of 50 nm, 61 g of diacetone alcohol (DAA) and 4.0 g of a coupling agent for dispersing the fine particles were mixed for twelve hours in a ball mill by using zirconia balls in a ball mill to prepare 100 g of a dispersant of ATO fine particles (solution D). The solution D, the dispersion A disclosed in Example 1 and silicon-based UV curing resin produced by Shin-Etsu Silicon Co., Ltd. were thoroughly mixed while diluting with ethanol to prepare a solution having the composition as shown by Example 20 in Table 1. 15 g of the thus prepared solution was dropped from a beaker onto a sheet of soda-glass employed as a substrate, measuring 200 mm square by 3 mm thick and rotating at a speed of 200 rpm and after five minutes, its rotation was stopped. The resultant substrate was heated at 100°C for 2 minutes in an electric furnace to evaporate the solvent and, then, subjected to a ultraviolet irradiation for two minutes by using a high-pressure mercury lamp to obtain an aimed film. The optical characteristics of the thus obtained film are shown in Table 1.

COMPARATIVE EXAMPLE 2

[0061] A coating liquid containing LaB₆ dispersed in a silicate solution was prepared by using LaB₆ having an average particle diameter of 137 nm and otherwise repeating Example 1, and was used to form a film by repeating Example 1. The particles were, however, so large that the film had a haze of 28%, lacked transparency, and was slightly greenish gray. Moreover, it had a difference of as small as 12% between its maximum and minimum transmittances, and was considered difficult to use for cutting off heat rays.

35 COMPARATIVE EXAMPLE 3

[0062] A coating liquid containing ITO dispersed in a silicate solution was prepared by using ultrafine particles of ITO having an average diameter of 22 nm and otherwise repeating Example 1, and was used to form a film by repeating Example 1. The film, however, showed a transmittance of over 90% from visible light to an infrared wavelength of 1500 nm, and these results taught that a film containing only 1% of ITO was useless for cutting off near infrared radiation.

[Table 1]

	EXAMPLE	Dispersed fine particles	Binder	Overcoat layer	Visible light transmittance(%)	maximum transmittance wavelength(nm)	minimum transmittance wavelength(nm)	Difference between maximum and minimum(points)	Surface resistivity(Ω/\square)
5	1	LaB ₆	Silicate	None	54.6	580	1055	38	3.2×10^{11}
10	2	LaB ₆	Silicate	Silicate	56.2	584	1060	34	9.5×10^{11}
15	3	LaB ₆	Epoxy resin	None	48.1	575	1039	23	6.5×10^{11}
20	4	LaB ₆	Cold-setting Silicate	None	51.3	587	1104	32	1.4×10^{12}
25	5	LaB ₆	Silazane	None	51.9	576	1033	42	2.2×10^{11}
30	6	PrB ₆	Silicate	None	55.7	594	1108	26	6.4×10^{11}
35	7	NdB ₆	Silicate	None	62.1	573	1060	42.8	4.5×10^{11}
40	8	CeB ₆	Silicate	None	68.5	575	1075	49.5	8.4×10^{10}
45	9	GdB ₆	Silicate	None	68.1	596	1104	32.3	2.6×10^{11}
50	10	TbB ₆	Silicate	None	54.0	589	1109	31.2	3.3×10^{11}
55	11	DyB ₆	Silicate	None	65.8	578	1095	33.0	8.7×10^{10}
60	12	HoB ₆	Silicate	None	48.7	588	1094	29.4	7.8×10^{11}
65	13	YB ₆	Silicate	None	78.3	589	1102	47.5	4.5×10^{11}
70	14	SmB ₆	Silicate	None	56.5	591	1084	33.7	7.3×10^{11}
75	15	EuB ₆	Silicate	None	54.2	587	1098	30.5	2.2×10^{12}
80	16	ErB ₆	Silicate	None	50.9	593	1115	33.3	9.2×10^{11}
85	17	TmB ₆	Silicate	None	46.0	586	1112	30.2	8.5×10^{11}
90	18	YbB ₆	Silicate	None	63.8	583	1091	31.2	4.3×10^{12}
95	19	LuB ₆	Silicate	None	64.3	593	1100	35.2	9.6×10^{11}
100	20	SrB ₆	Silicate	None	77.2	605	1125	25.6	9.9×10^{12}
105	21	CaB ₆	Silicate	None	78.2	599	1116	24.3	8.8×10^{12}
110	22	TiB ₂	Silicate	None	63.4	493	986	29	2.5×10^{11}
115	23	TaB ₂	Silicate	None	52.8	608	1104	24	2.5×10^{12}
120	24	MoB	Silicate	None	49.6	456	978	28	8.4×10^{10}
125	25	W ₂ B ₅	Silicate	None	58.2	591	1071	23	5.3×10^{11}
130	26	LaB ₆ +RuO ₂	Silicate	None	64.3	583	1034	31	7.8×10^{11}
135	27	LaB ₆ +IrO ₂	Silicate	None	62.5	589	1046	30	8.6×10^{11}
140	28	CeB ₆ +ITO	Silicate	None	70.5	570	1800	66.0	7.6×10^{11}
145	29	CeB ₆ +ATO	Silicate	None	67.3	573	1800	62.0	9.8×10^{11}
150	COMPARATIVE EXAMPLE 1	-	-	None	38.8	640	1000	8	83
155	COMPARATIVE EXAMPLE 2	LaB ₆	Silicate	None	61.7	572	1024	12	5.8×10^{11}
160	COMPARATIVE EXAMPLE 3	ITO	Silicate	None	93.5	-	-	-	8.3×10^3

50

Claims

- 55 1. A coating liquid for forming a film for cutting off heat rays which comprises a dispersion containing fine particles of boride having an average diameter of 100 nm at most.
2. A coating liquid according to claim 1, wherein said boride is selected from the group consisting of lanthanum boride

(LaB₆), praseodymium boride (PrB₆), neodymium boride (NdB₆), cerium boride (CeB₆), gadolinium boride (GdB₆), terbium boride (TbB₆), dysprosium boride (DyB₆), holmium boride (HoB₆), yttrium boride (YB₆), samarium boride (SmB₆), europium boride (EuB₆), erbium boride (ErB₆), thulium boride (TmB₆), ytterbium boride (YbB₆), lutetium boride (LuB₆), strontium boride (SrB₆), calcium boride (CaB₆), titanium boride (TiB₂), zirconium boride (ZrB₂), hafnium boride (HfB₂), vanadium boride (VB₂), tantalum boride (TaB₂), chromium borides (CrB and CrB₂), molybdenum borides (MoB₂, Mo₂B₅ and MoB) and tungsten boride (W₂B₅), said particles having an average diameter of 100 nm at most.

- 5 3. A coating liquid according to claim 1 or claim 2, wherein said dispersion further contains fine particles of at least one or ruthenium and iridium oxides having an average diameter of 100 nm at most.
- 10 4. A coating liquid according to claim 3, wherein said ruthenium oxide is selected from the group consisting of ruthenium dioxide (RuO₂), lead ruthenate (Pb₂Ru₂O_{6.5}) and bismuth ruthenate (Bi₂Ru₂O₇).
- 15 5. A coating liquid according any preceding claim, wherein said dispersion further contains at least one oxide selected from the group consisting of indium tin oxide (ITO), antimony tin oxide (ATO), and aluminium zinc oxide (AZO).
- 20 6. A coating liquid according to any preceding claim, wherein said dispersion further contains at least one compound selected from the group consisting of alkoxides of silicon, zirconium, titanium and aluminium, and partially polymerized products of said alkoxides.
- 25 7. A coating liquid according to claim 6, wherein said compound in terms of oxide occupies at most 40% by weight of the liquid.
- 30 8. A coating liquid according to any preceding claim, wherein said dispersion contains a resinous binder.
- 35 9. A coating liquid according to claim 8, wherein said dispersion is prepared by dispersing said particles in said binder uniformly by using a bead, ball or sand mill, or an ultrasonic disperser.
- 40 10. A film formed on a substrate for cutting off heat rays by coating the substrate with a coating liquid and heating it, the coating liquid comprising a dispersion prepared by dispersing fine particles of at least one boride selected from the group consisting of LaB₆, PrB₆, NdB₆, CeB₆, GdB₆, TbB₆, DyB₆, HoB₆, YB₆, SmB₆, EuB₆, ErB₆, TmB₆, YbB₆, LuB₆, SrB₆, CaB₆, TiB₂, ZrB₂, HfB₂, VB₂, TaB₂, CrB, CrB₂, MoB₂, Mo₂B₅, MoB and W₂B₅ in a resinous binder, or a binder containing one or more metal oxides selected from the group consisting of silicon, zirconium, titanium and aluminium oxides.
- 45 11. A film according to claim 10, wherein said dispersion further contains fine particles of at least one of ruthenium and iridium oxides.
- 50 12. A film according to claim 10 or claim 11, wherein said dispersion further contains fine particles of at least one of indium tin oxide and antimony tin oxide.
- 55 13. A film according to any of claims 10 to 12, the film having a maximum transmittance of light at a wavelength of 400 to 700 nm and a minimum transmittance of a wavelength of 700 to 1800 nm with a difference of at least 15% therebetween.
- 60 14. A film according to any of claims 10 to 13, the film having a surface resistivity of at least 10⁶ Ω/□.
- 65 15. A film according to any of claims 10 to 12, wherein said dispersion is prepared by dispersing said particles in said binder uniformly by using a bead, ball or sand mill, or an ultrasonic disperser.
- 70 16. A film according to any of claims 10 to 12, wherein said substrate is spin, spray, dip, roll or flow coated, or screen printed with said coating liquid.
- 75 17. A multilayer film for cutting off heat rays comprising a first layer formed on a substrate by coating it with a coating liquid and heating it, the coating liquid comprising a dispersion of fine particles of a boride of one or more metals selected from the group consisting of LaB₆, PrB₆, NdB₆, CeB₆, GdB₆, TbB₆, DyB₆, HoB₆, YB₆, SmB₆, EuB₆, ErB₆, TmB₆, YbB₆, LuB₆, SrB₆, CaB₆, TiB₂, ZrB₂, HfB₂, VB₂, TaB₂, CrB, CrB₂, MoB₂, Mo₂B₅, MoB and W₂B₅, and a resin

layer or an oxide later formed on said first layer, said oxide layer containing at least one metal oxide selected from the group consisting of silicon, zirconium, titanium and aluminium oxides.

18. A multilayer film according to claim 17, wherein said dispersion further contains fine particles of at least one of ruthenium and iridium oxides.
5
19. A multilayer film according to claim 17 or claim 18, said film having a maximum transmittance of light at a wavelength of 400 to 700 nm and a minimum transmittance at a wavelength of 700 to 1800 nm with a difference of at least 15% therebetween.
10
20. A multilayer film according to any of claims 17 to 19, said film having a surface resistivity of at least $10^6 \Omega/\square$.
21. A multilayer film according to claim 17 or claim 18, wherein said dispersion is prepared by dispersing said particles uniformly by using a bead, ball or sand mill, or an ultrasonic disperser.
15
22. A multilayer film according to claim 17 or claim 18, wherein said substrate is spin, spray, dip, roll or flow coated, or screen printed with said coating liquid.

20

25

30

35

40

45

50

55

Fig. 1

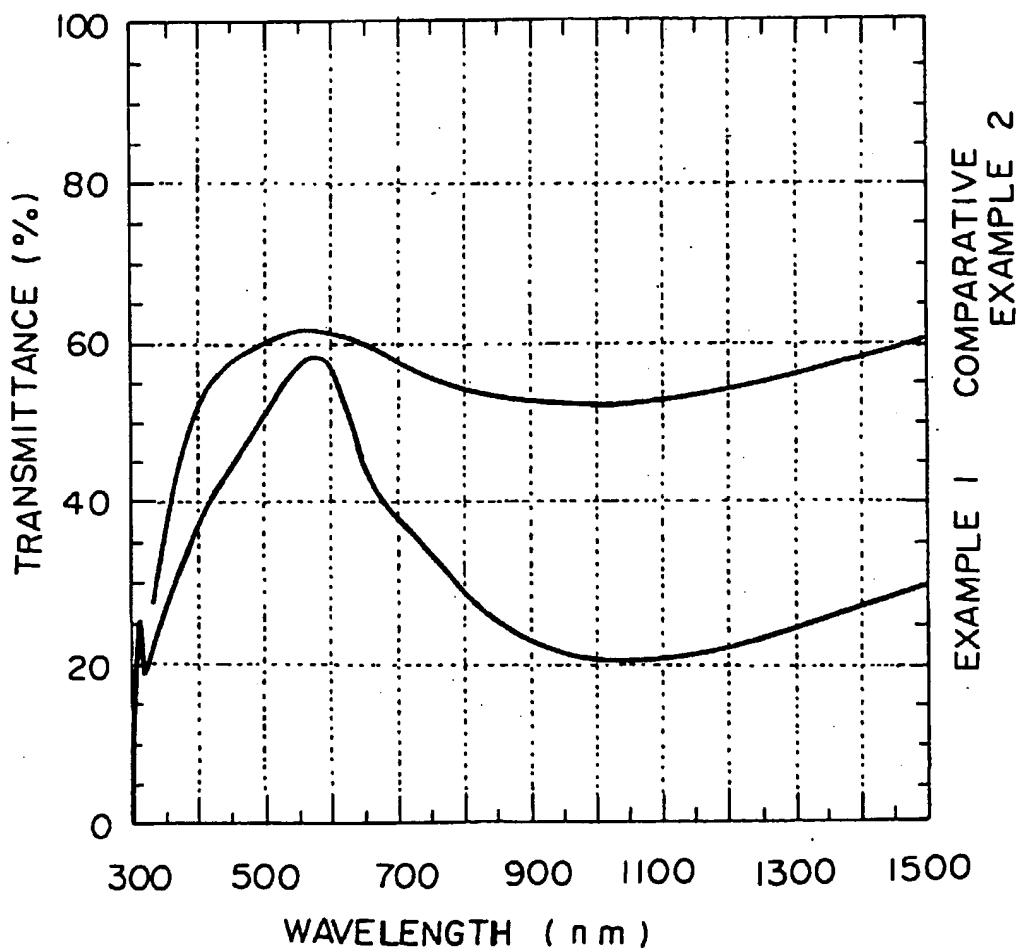
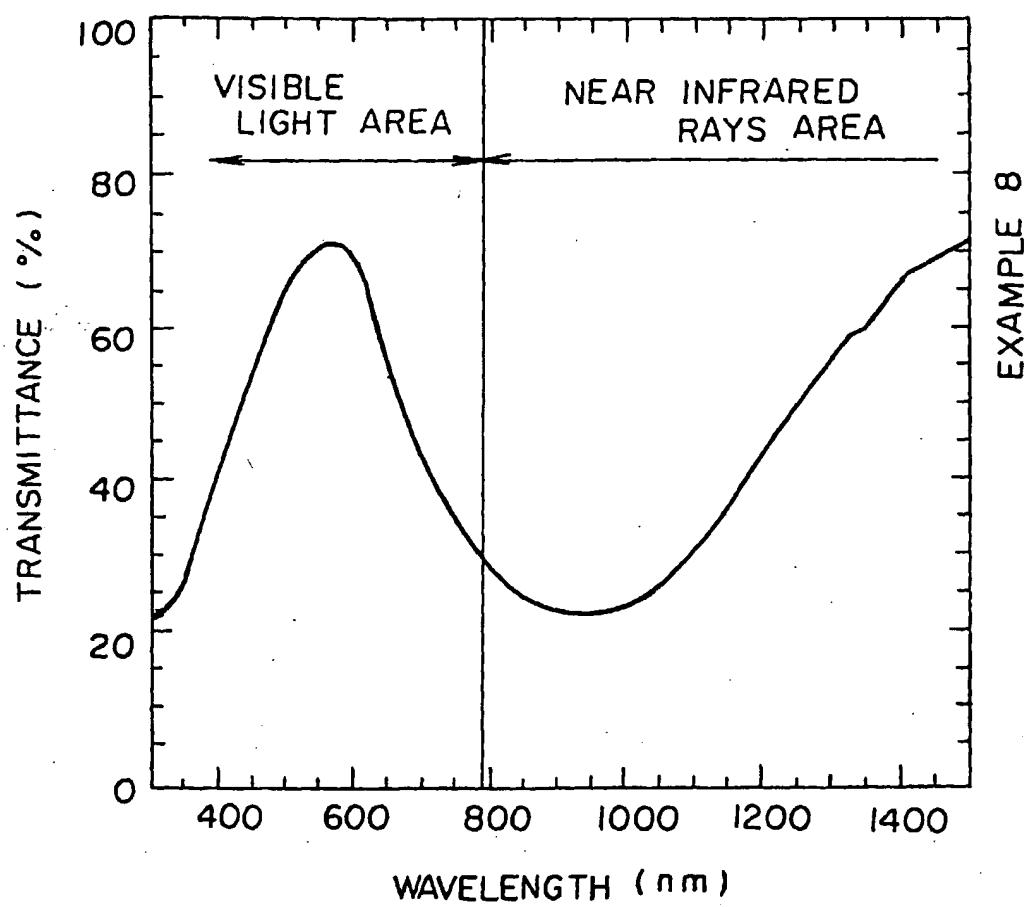


Fig. 2





European Patent
Office

EUROPEAN SEARCH REPORT

Application Number
EP 98 30 8885

DOCUMENTS CONSIDERED TO BE RELEVANT			
Category	Citation of document with indication, where appropriate, of relevant passages	Relevant to claim	CLASSIFICATION OF THE APPLICATION (Int.Cl.6)
E	EP 0 905 100 A (SUMITOMO METAL MINING CO) 31 March 1999 (1999-03-31) * example 11 * * tables * * claims * ---	1,2, 6-10, 13-17, 19-22	C03C17/00 C03C17/22 C03C17/34 C03C17/42 C09D1/00 C09D5/00
P,A	WO 98 46539 A (IWATA MISAQ ;KATO SHINJI (JP); FUKUI TAKAMITSU (JP); HORIMI KAZUHI) 22 October 1998 (1998-10-22) * page 2, line 18 * * page 4, line 8 - line 9 * * page 4, line 26 - line 42 * * page 7, line 1 - line 2 * * page 13, line 12 - line 13 * * page 12, line 17 * * claims 19,20 * -& EP 0 911 299 A ---	1	
E			
A	EP 0 454 349 A (HUGHES AIRCRAFT CO) 30 October 1991 (1991-10-30) * page 5, line 48 - line 51 * * claim 2 * ---	1,2,10, 17	TECHNICAL FIELDS SEARCHED (Int.Cl.6) C03C C09D
A	EP 0 779 343 A (SUMITOMO METAL MINING CO) 18 June 1997 (1997-06-18) * page 2, line 29 - line 32 * * page 2, line 46 - page 3, line 25 * * page 3, line 45 - page 4, line 13 * * page 4, line 25 - line 45 * * examples * * tables * * claims * ---	3-9,11, 12,15, 16,18, 21,22	
		-/-	
<p>The present search report has been drawn up for all claims</p>			
Place of search	Date of completion of the search	Examiner	
MUNICH	23 July 1999	Somann, K	
CATEGORY OF CITED DOCUMENTS		T : theory or principle underlying the invention E : earlier patent document, but published on, or after the filing date D : document cited in the application L : document cited for other reasons & : member of the same patent family, corresponding document	
<small>EPO FORM 1503/02 (PAC01)</small>			
<small>X : particularly relevant if taken alone Y : particularly relevant if combined with another document of the same category A : technological background C : non-written disclosure P : intermediate document</small>			



European Patent
Office

EUROPEAN SEARCH REPORT

Application Number

EP 98 30 8885

DOCUMENTS CONSIDERED TO BE RELEVANT									
Category	Citation of document with indication, where appropriate, of relevant passages	Relevant to claim	CLASSIFICATION OF THE APPLICATION (Int.Cl.6)						
A	US 3 885 855 A (GROSS DANIEL M) 27 May 1975 (1975-05-27) * column 3, line 58 * * column 3, line 64 * * column 5, line 40 * * column 5, line 49 - line 51 * * column 6, line 24 - line 36 * * claim 1 *	1,2,10, 17							
P,A	PATENT ABSTRACTS OF JAPAN vol. 098, no. 009, 31 July 1998 (1998-07-31) & JP 10 101375 A (SUMITOMO METAL MINING CO LTD), 21 April 1998 (1998-04-21) * abstract *	3-9,11, 12,15, 16,18, 21,22							
A	PATENT ABSTRACTS OF JAPAN vol. 098, no. 006, 30 April 1998 (1998-04-30) & JP 10 036774 A (SUMITOMO METAL MINING CO LTD), 10 February 1998 (1998-02-10) * abstract *	3,4,11, 17							
A	PATENT ABSTRACTS OF JAPAN vol. 017, no. 577 (C-1122), 20 October 1993 (1993-10-20) & JP 05 170485 A (ASAHI GLASS CO LTD), 9 July 1993 (1993-07-09) * abstract *	1,2,10, 17	TECHNICAL FIELDS SEARCHED (Int.Cl.6)						
<p>The present search report has been drawn up for all claims</p> <table border="1" style="width: 100%; border-collapse: collapse;"> <tr> <td style="width: 33%;">Place of search</td> <td style="width: 33%;">Date of completion of the search</td> <td style="width: 34%;">Examiner</td> </tr> <tr> <td>MUNICH</td> <td>23 July 1999</td> <td>Somann, K</td> </tr> </table>				Place of search	Date of completion of the search	Examiner	MUNICH	23 July 1999	Somann, K
Place of search	Date of completion of the search	Examiner							
MUNICH	23 July 1999	Somann, K							
CATEGORY OF CITED DOCUMENTS		T : theory or principle underlying the invention E : earlier patent document, but published on, or after the filing date D : document cited in the application L : document cited for other reasons & : member of the same patent family, corresponding document							
<small>EPO FORM 1503/02 (P04C01)</small>									

**ANNEX TO THE EUROPEAN SEARCH REPORT
ON EUROPEAN PATENT APPLICATION NO.**

EP 98 30 8885

This annex lists the patent family members relating to the patent documents cited in the above-mentioned European search report.
 The members are as contained in the European Patent Office EDP file on
 The European Patent Office is in no way liable for these particulars which are merely given for the purpose of information.

23-07-1999

Patent document cited in search report		Publication date		Patent family member(s)	Publication date
EP 0905100	A	31-03-1999	AU	8716998 A	22-04-1999
WO 9846539	A	22-10-1998	JP	10287443 A	27-10-1998
			CA	2257798 A	22-10-1998
			EP	0911299 A	28-04-1999
EP 0454349	A	30-10-1991	AU	635629 B	25-03-1993
			AU	7431491 A	07-11-1991
			CA	2039109 A	24-10-1991
			JP	4230703 A	19-08-1992
			JP	7101242 B	01-11-1995
			KR	9400718 B	28-01-1994
			US	5405680 A	11-04-1995
EP 0779343	A	18-06-1997	JP	9156963 A	17-06-1997
			JP	9286637 A	04-11-1997
			JP	9302284 A	25-11-1997
			US	5840364 A	24-11-1998
US 3885855	A	27-05-1975	BE	787599 A	16-02-1973
			CH	544043 A	28-12-1973
			DE	2240409 A	01-03-1973
			FR	2149487 A	30-03-1973
			GB	1397316 A	11-06-1975
			JP	48029815 A	20-04-1973
JP 10101375	A	21-04-1998		NONE	
JP 10036774	A	10-02-1998		NONE	
JP 05170485	A	09-07-1993		NONE	